CHAPTER 3 CHAPTER ASSESSMENT

MATTER AND CHANGE

Reviewing Vocabulary

March the definition in Column A with the term in Column B.

Column A.

- The outermost electrons of an atom
- _____ 2. The center of an atom
 - 3. Hot, highly ionized, electrically conducting gas
 - The smallest particle of an element that retains that element's characteristics
 - 5. The change from a gas to a liquid
- _____ 6. A homogeneous mixture
 - 7. The change of one or more substances into other substances
 - The attractive force between two ions of opposite charge
 - A tiny particle with a negative electrical charge
 - ___ 10. The average of the mass numbers of an element's isotopes
- The spontaneous process through which unstable nuclei emit radiation

Column B

- as, agoon
- b. atomic mass
- e. chemical reaction
- d. condensation
- e. electron
- f. ionic bond
- g. solution
- h. nucleus
- i. plasma
- valence electrons
- k. radioactive decay

In the space at the left, write the word or phrase in parentheses that makes the statement correct.

- The combined number of protons and neutrons is the (atomic number, mass number).
- An (energy level, isotope) represents the area in an atom where an electron is most likely to be found.
- 14. (An ionic bond, A covalent bond) is the attraction of two atoms for a shared pair of electrons that holds the atoms together.
- An atom that gains or loses an electron is a charged particle called (an ion, a molecule).
- (An element, A compound) is a substance that cannot be broken down into simpler substances.
- A tiny particle that has mass and a positive electrical charge is a (neutron, proton).
- (Acids, Bases) are solutions characterized by the formation of hydroxide ions (OH⁺).
- The process of changing from a liquid into a gas is called (evaporation, sublimation).

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