

**CHEMISTRY TEST**

NAME: \_\_\_\_\_

**INTRODUCTORY CHEMISTRY CONCEPTS**

DATE: \_\_\_\_\_

**Directions:** For each of the following questions, choose the number that **best** answers the question and place it on your answer sheet.

**Directions:** For each of the following questions, all major steps must be shown to receive full credit.

**Introductory Chemistry Concepts****Level 1****Multiple Choice**

1. A chemical change always results characteristically in a change of

- 1) state;                      2) form;                      3) identity;                      4) size;                      5) shape.

ANS: 3

2. Which of the following is a chemical change?

- 1) evaporation of water                      2) distillation of water                      3) freezing of water  
4) electrolysis of water

ANS: 4

3. Chemical changes always involve    1) the absorption of energy;                      2) the liberation of energy;  
3) either the absorption or liberation of energy;                      4) changes of state.

ANS: 3

4. One physical property of sugar is its    1) acidity;                      2) reactivity;

- 3) flammability;                      4) plasma qualities;                      5) solubility.

ANS: 5

5. The majority of chemical changes are    1) autothermic;                      2) endothermic;

- 3) isothermic;                      4) nonthermic;                      5) exothermic.

ANS: 5

6. The statement which describes a **physical** property of the element oxygen is

- 1) oxygen supports the burning of paper;                      2) oxygen is needed for the rusting of iron;  
3) oxygen must be present for food to decay;                      4) oxygen has a density of 1.43 grams per liter;  
5) oxygen is a component of water.

ANS: 4

7. In any chemical reaction energy is    1) always absorbed;                      2) always released;

- 3) released as heat;                      4) usually not involved;                      5) either absorbed or released.

ANS: 5

8. Which of the following is **not** a chemical change?

- 1) the dissolving of zinc by HCl                      2) the exploding of gasoline                      3) the corroding of a copper roof  
4) the boiling of water

ANS: 4

9. One chemical property of a substance is a description of its    1) odor;                      2) density;

- 3) combustibility;                      4) solubility;                      5) none of these.

ANS: 3

10. A new substance is formed as a result of a    1) physical change;                      2) chemical

# Chemistry Matter Change Section Assessment Answers

**James E. Brady, John R. Holum**



## **Chemistry Matter Change Section Assessment Answers:**

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