

# Chemistry JAMB Questions and Answers 2023

## [Numbers 1 – 10]

1. Which of the following is a mixture?

- A. Sodium chloride
- B. Sea water
- C. Iron filings
- D. Granulated sugar

2. Two elements, X and Y, have atomic numbers 8 and 13 respectively. The formula for the possible compound found between X and Y is

- A.  $Y_2X_3$
- B.  $XY_2$
- C.  $X_2Y_3$
- D.  $X_3Y_2$

3. 3 g of a mixture of  $CaO$  and  $CaCO_3$  was heated to a constant mass. If 0.44 g of  $CO_2$  was liberated, calculate the percentage of  $CaO$  in the mixture.

- A. 33.3%
- B. 50%
- C. 66.67%
- D. 25%

4. An alkanoic acid has a molecular mass of 88. Its molecular formula is

- A.  $C_4H_9COOH$
- B.  $C_5H_{11}COOH$
- C.  $C_3H_5COOH$
- D.  $C_3H_7COOH$

5. If the rate of diffusion of oxygen is taken as  $1\text{ cm}^3\text{ s}^{-1}$ , what will be the rate of diffusion of methane whose relative molecular mass is 16?

- A. 2.0
- B. 1.8
- C. 1.4
- D. 1.0

6. An increase in temperature causes an increase in the pressure of a gas in a fixed volume due to an increase in the

- A. Number of molecules of the gas
- B. Density of the gas molecules
- C. Number of collisions between the gas molecules
- D. Number of collisions between the gas molecules and the walls of the container

7. In electrolysis, the chemical reaction which takes place at the anode is

- A. dissociation
- B. hydrolysis
- C. oxidation
- D. reduction

8. In the electrolysis of brine, it is essential to prevent the mixing of the products because

- A. sodium and chlorine readily combine
- B. chlorine gives a green coloration
- C. chlorine readily recombines with sodium hydroxide
- D. sodium hydroxide forms a carbonate in the presence of air and chlorine

9. In what way is equilibrium constant for the forward reaction related to that of the reverse reaction?

- A. The two equilibrium constants are identical
- B. The product of the two is always greater than one
- C. The product of the two is expected to be one
- D. The addition of the two is expected to be one

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